



Synthesis and Characterization of Fluoropropyl POSS



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AFRL/PRSP

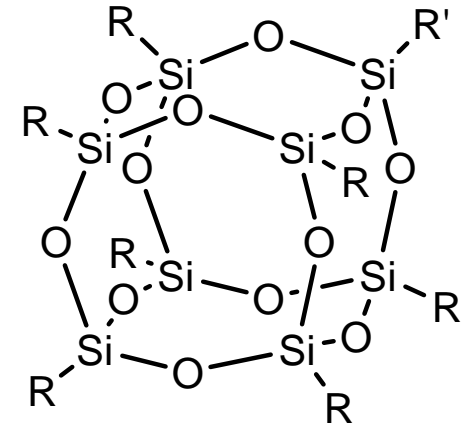
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Outline



- **POSS**
- **FluoroPOSS Synthesis**
- **Fluoropropyl POSS Synthesis**
- **Conclusion**

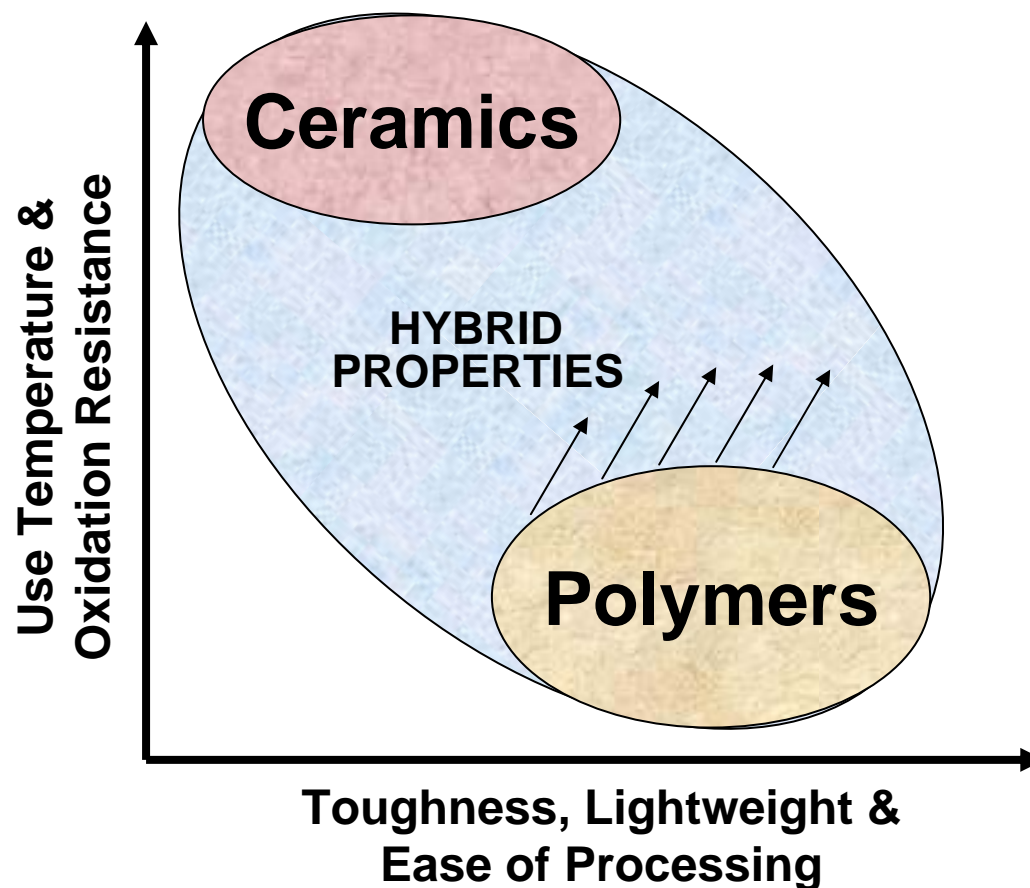




Hybrid inorganic/organic polymers



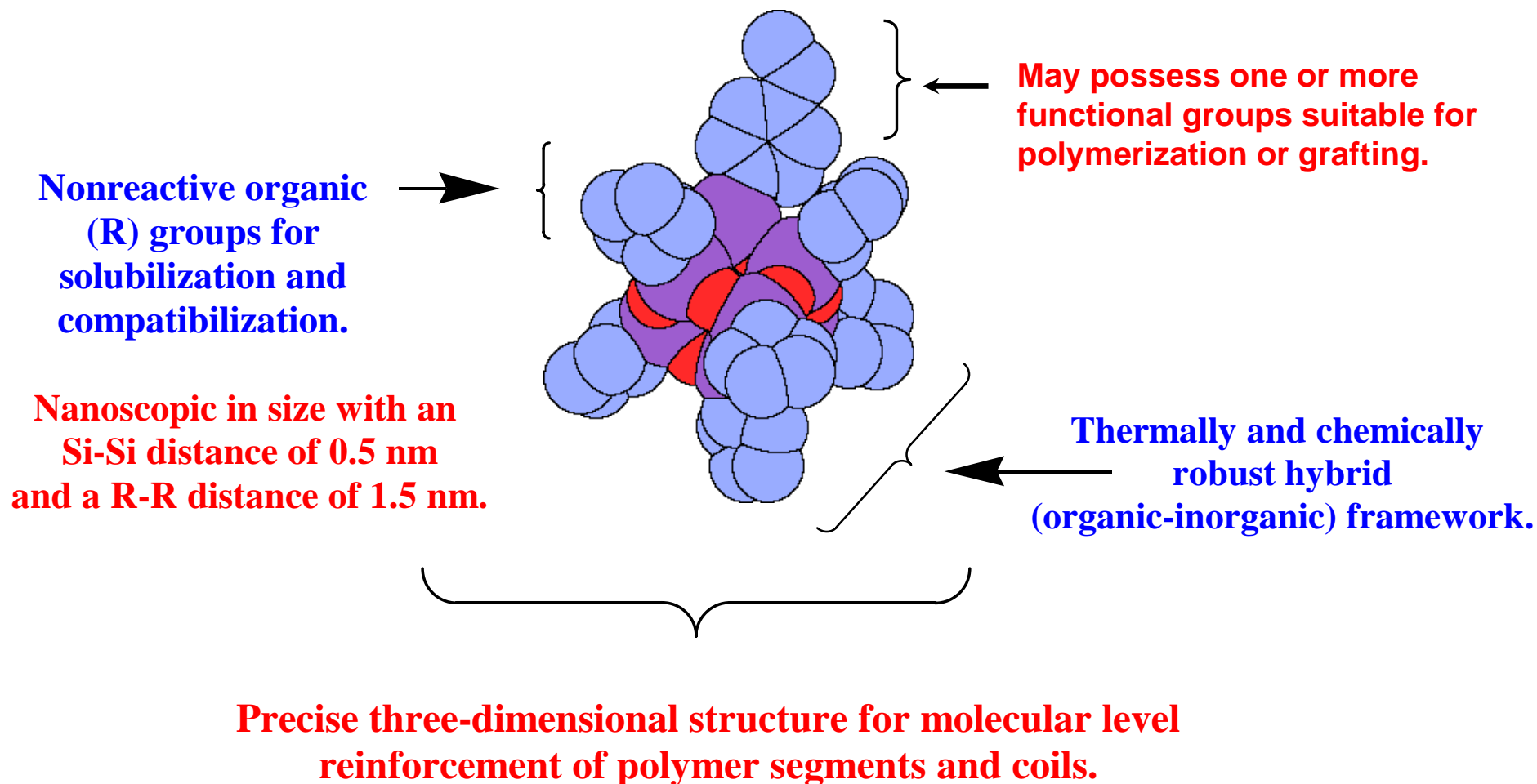
Goal: Develop High Performance Polymers that REDEFINE material properties



- **Hybrid plastics bridge the differences between ceramics and polymers**



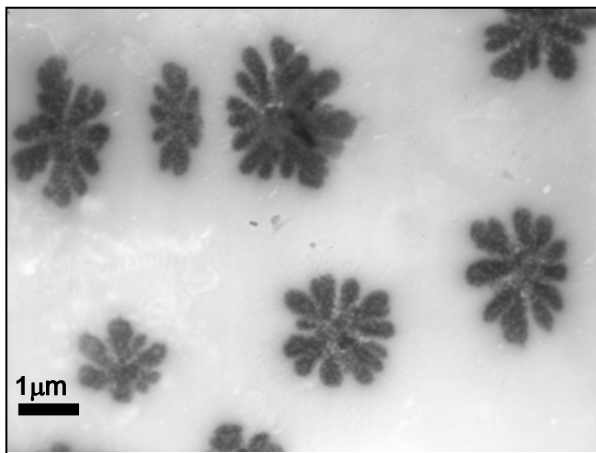
Anatomy of a POSS nanostructure



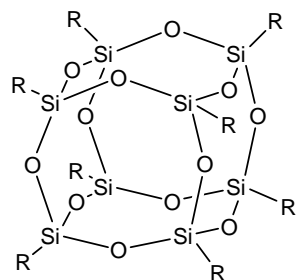


Importance of R groups: Affect compatibility with polymer matrix

50 Wt % POSS Blends in 2 Million MW PS

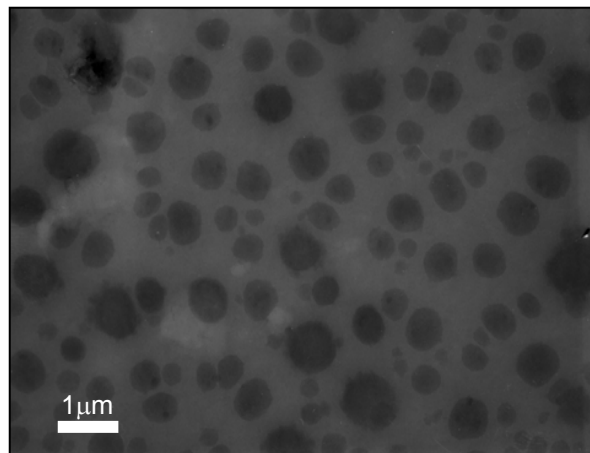


Domain Formation

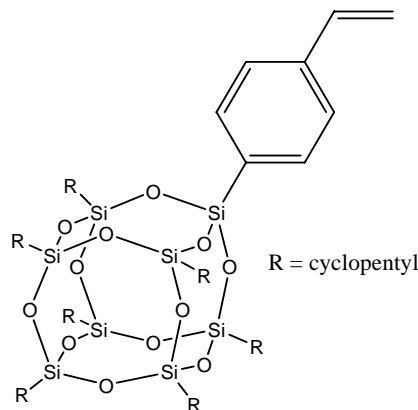


R = cyclopentyl

Cp₈T₈

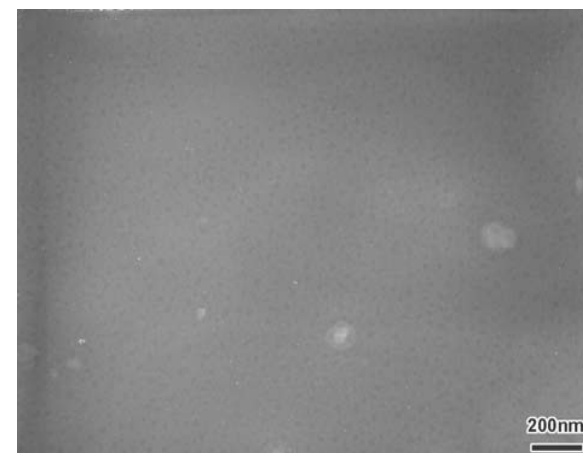


Partial Compatibility

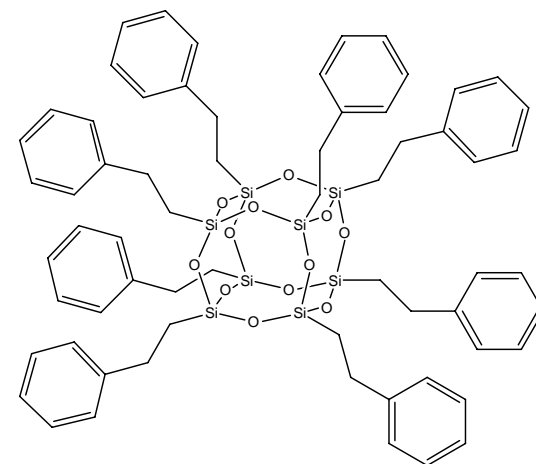


R = cyclopentyl

Cp₇T₈Styryl



Complete Compatibility



Phenethyl₈T₈



Fluoropolymer seals



Integrated High Payoff Rocket Propulsion Technology (IHPRPT) Program

- National program focused by OSD/DDR&E/AT and coordinated with AF, NASA, Navy, Army, and industry.
- Goal-oriented technology development program for rocket propulsion.
 - Double the performance of rocket propulsion systems over current state of the art.
 - Decrease the cost of access to space for commercial and military sectors.
- Fluoropolymers are used in many seal applications in liquid rocket engines
 - Polychlorotrifluoroethylene (PCTFE)
 - Fluorinated Ethylene/Propylene (FEP)
- Fluoropolymer research is designed to:
 - Improve Isp
 - Improve Thrust to Weight ratio
 - Reduce Support Costs
 - Lengthen Mean Time Between Removal





Targeted shortfalls



Mechanical Problems

- During operation, wear ring seals preventing propellant flow from bypassing the impeller blades in the SSME turbopump warp, causing an estimated 1.5 sec loss of delivered I_{sp} .

(I_{sp} and Thrust to Weight)

$$1.5 \text{ sec } I_{sp} \times 3 \text{ SSMEs} \times 1000 \text{ lb payload/sec } I_{sp} \times \$8,000/\text{lb payload} =$$

\$36,000,000 per launch

- Creep (cold flow of materials under load) causes the deformation of fluoropolymer seals in LREs, limiting their functional use time. (Mean Time Between Removal)

Surface Property Problems

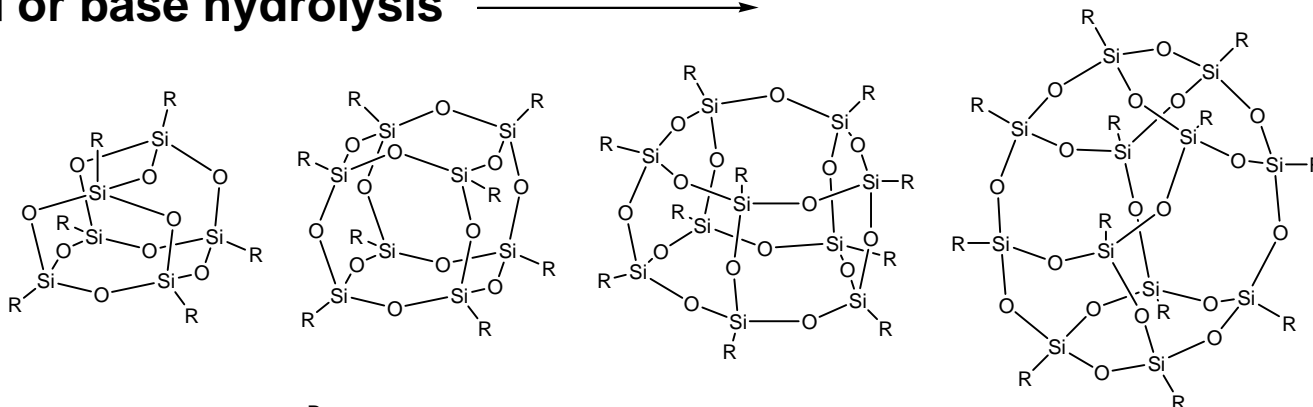
- Fuel wetting of fluoropolymer seals allows fuel to leak past seals. A costly nitrogen purge is necessary to prevent fuel contact with oxidizer. (Reduce Support Costs and Thrust to Weight)



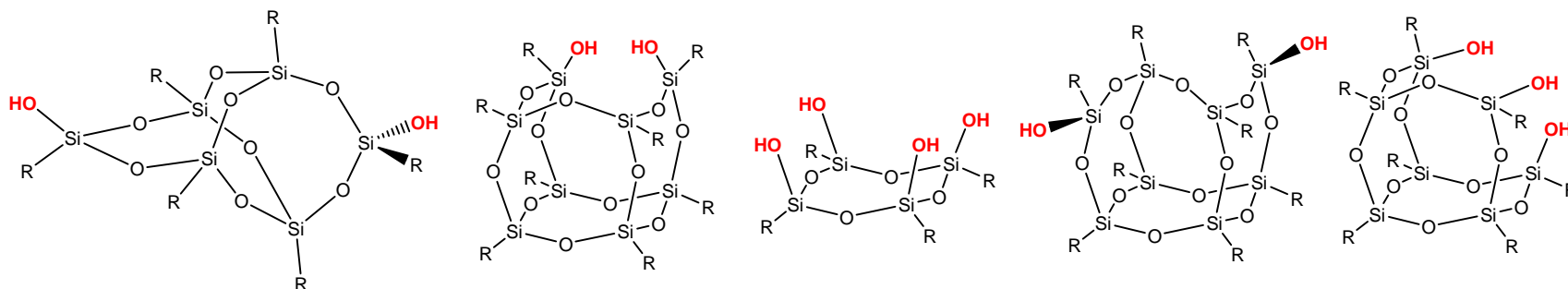
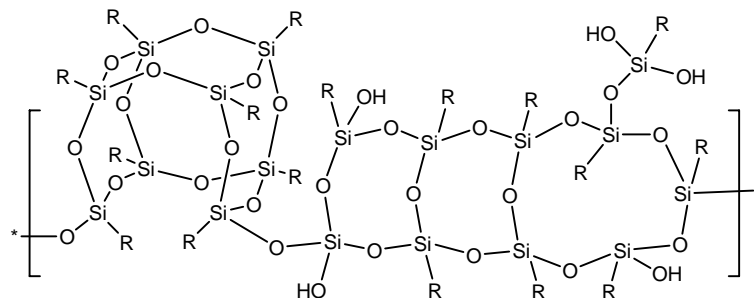
POSS synthesis

RSiX_3 acid or base hydrolysis \longrightarrow

Blendables



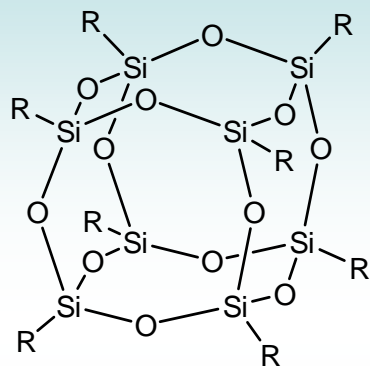
Resin



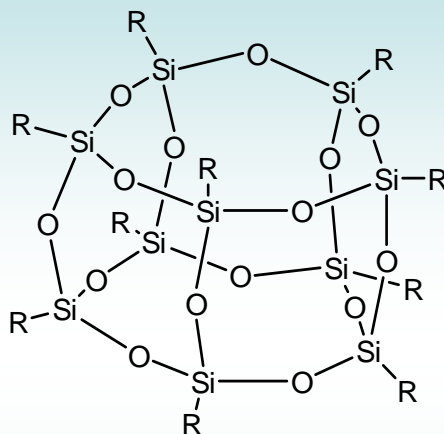
Incompletely condensed cages



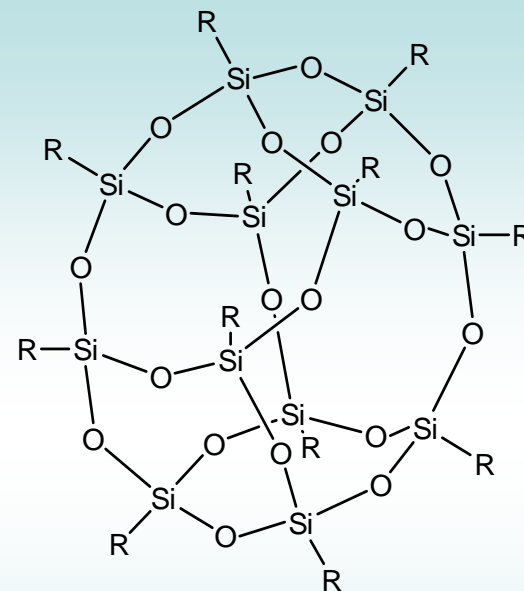
Typical isomers produced



T_8



T_{10}

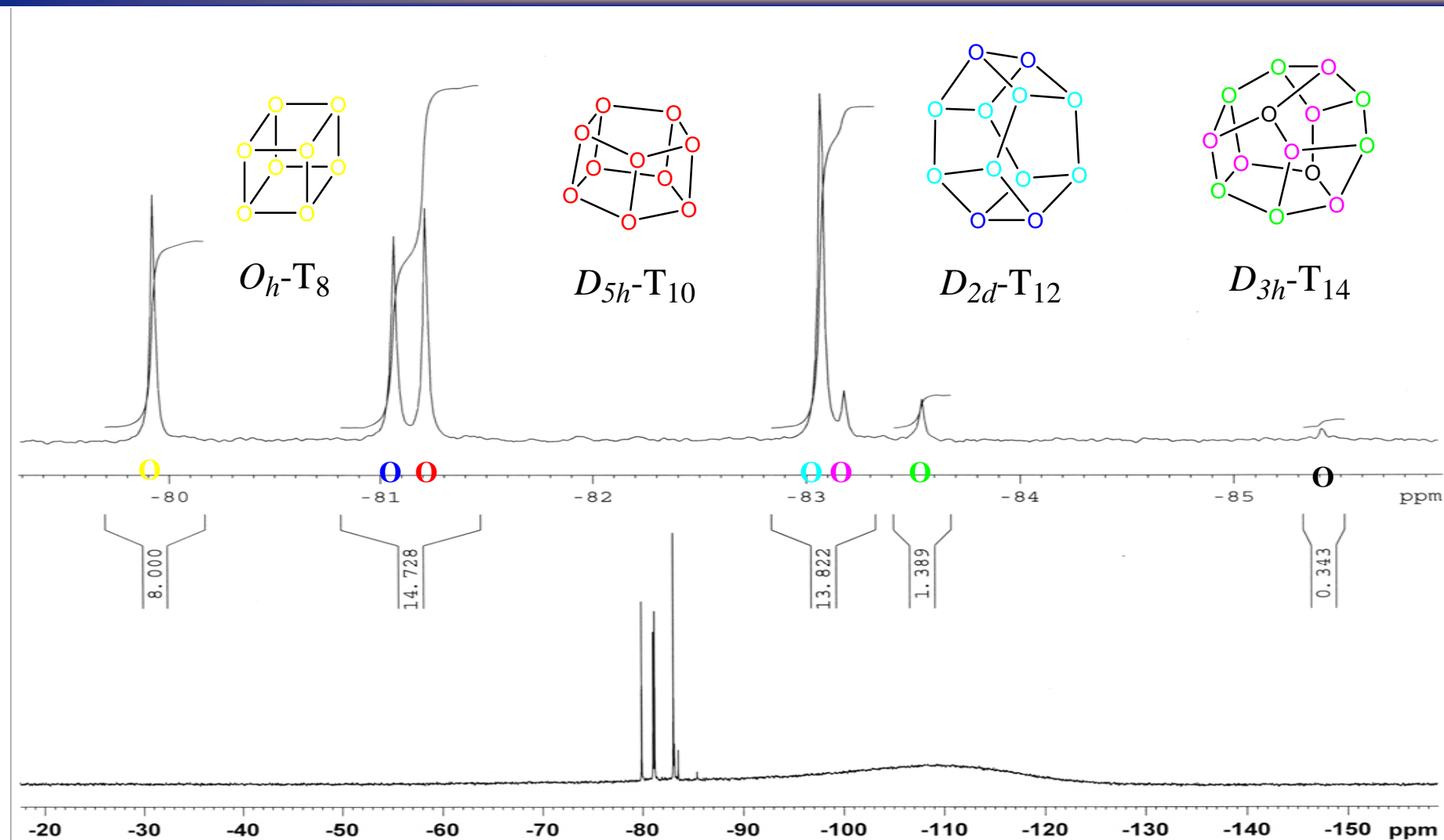


T_{12}

Most common compounds found in a cage mixture



^{29}Si NMR spectrum of cage mixture





²⁹Si NMR of Fluorohexyl cage mixture



Fluorohexyl Poss 02-08-05

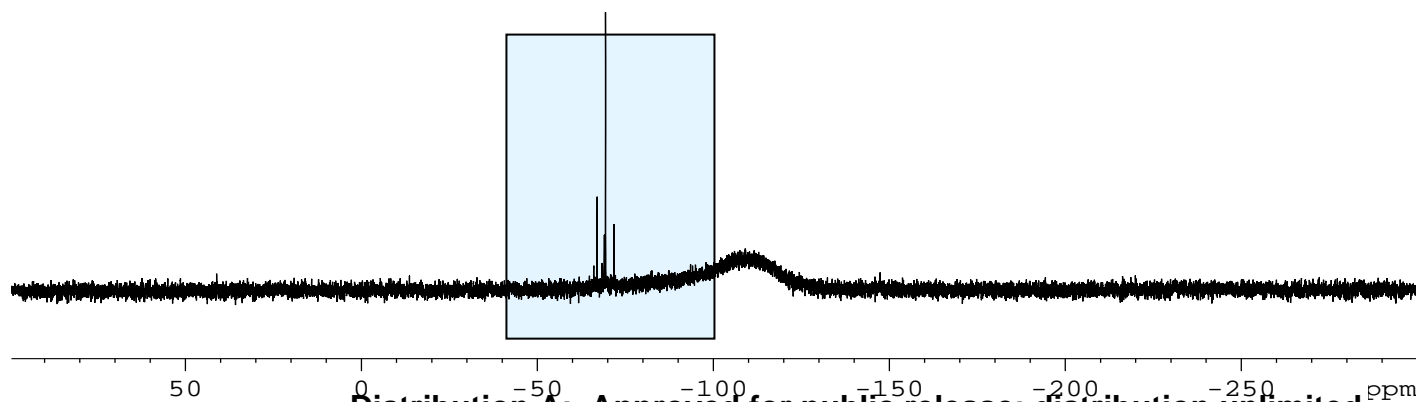
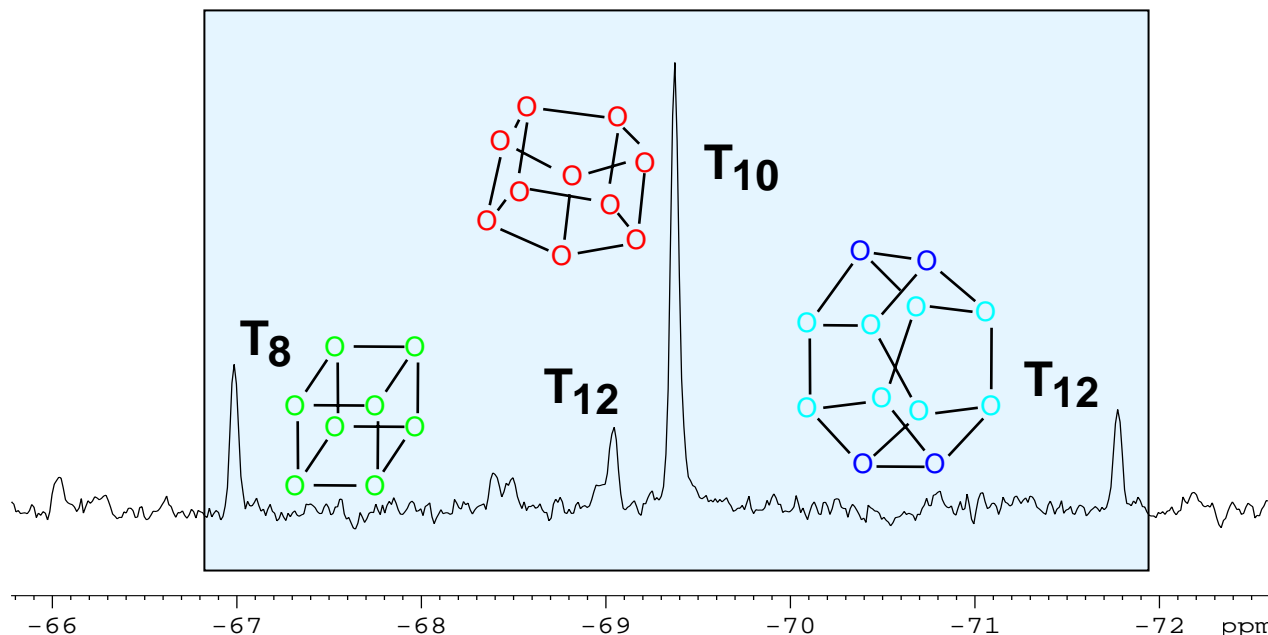
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EXPNO 29
PROCNO 1

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PULPROG zgig30
TD 65536
SOLVENT Acetone
NS 385
DS 4
SWH 23809.523 Hz
FIDRES 0.363304 Hz
AQ 1.3763061 sec
RG 11585.2
DW 21.000 usec
DE 6.00 usec
TE 300.0 K
D1 8.00000000 sec
d11 0.03000000 sec

===== CHANNEL f1 =====
NUC1 ²⁹Si
P1 10.00 usec
PL1 -3.00 dB
SFO1 59.6214106 MHz

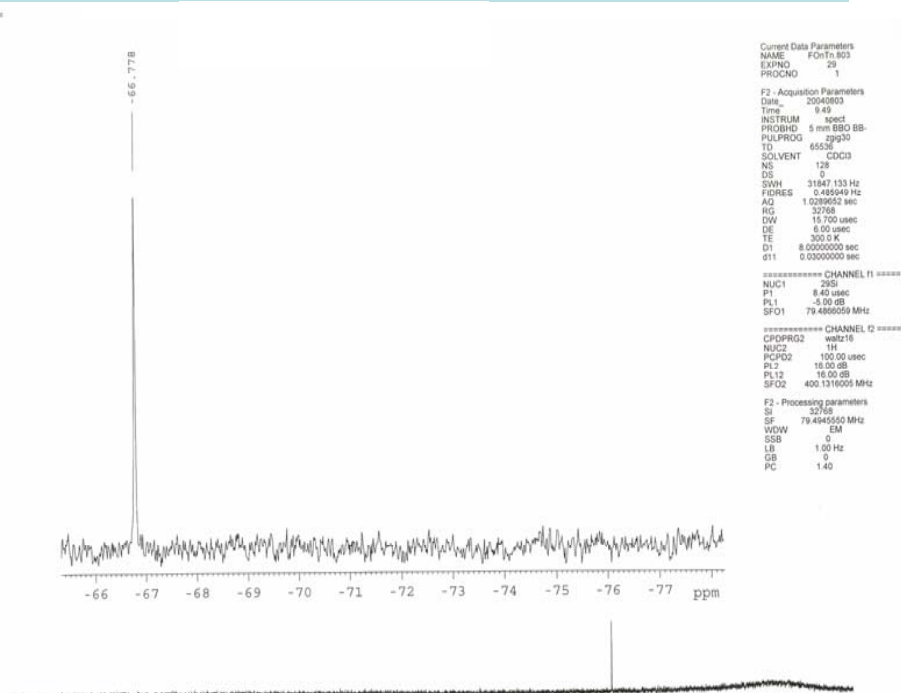
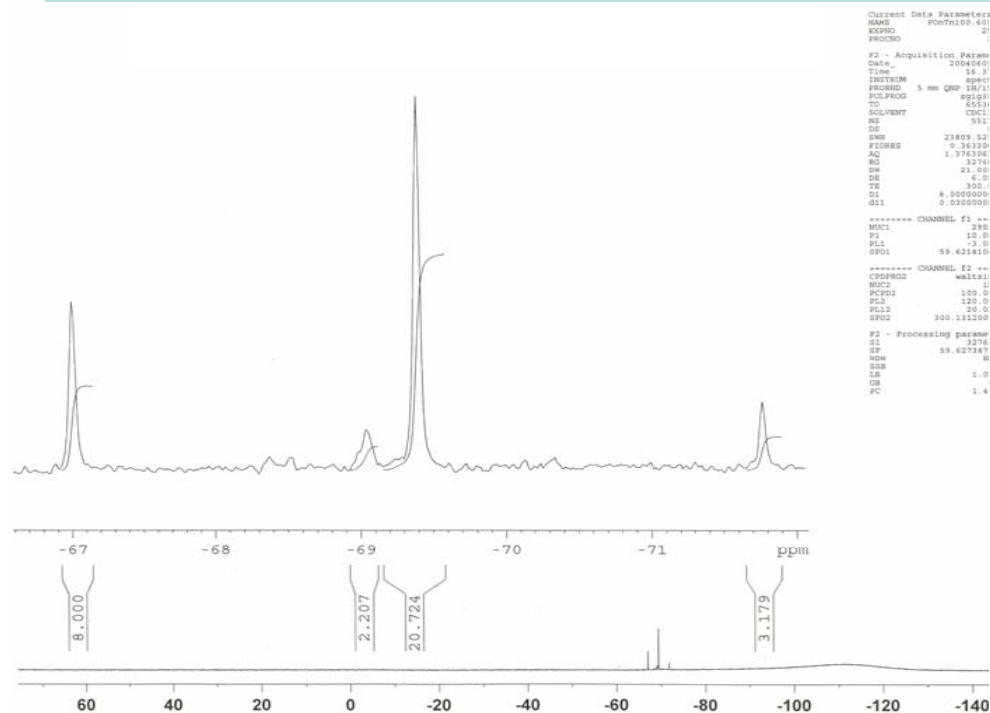
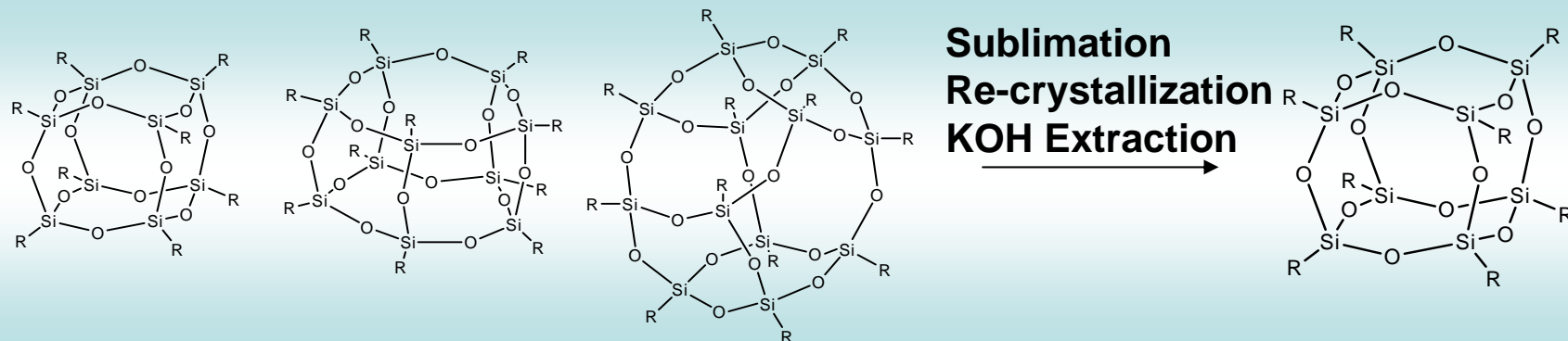
===== CHANNEL f2 =====
CPDPRG2 waltz16
NUC2 ¹H
PCPD2 100.00 usec
PL2 120.00 dB
PL12 20.00 dB
SFO2 300.1312005 MHz

F2 - Processing parameters
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SF 59.6273870 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40



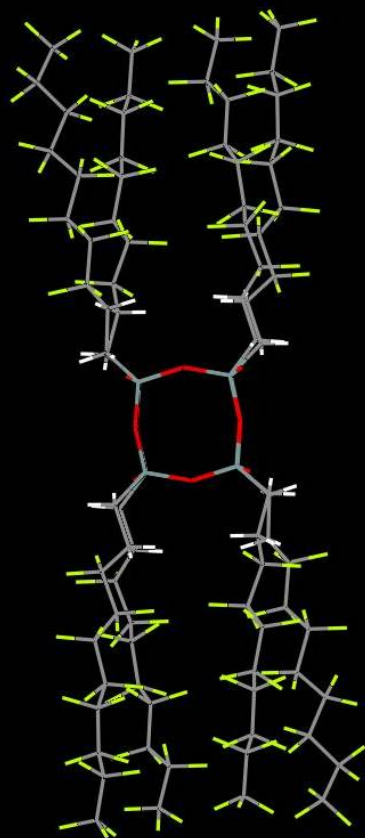


Redistribution reaction

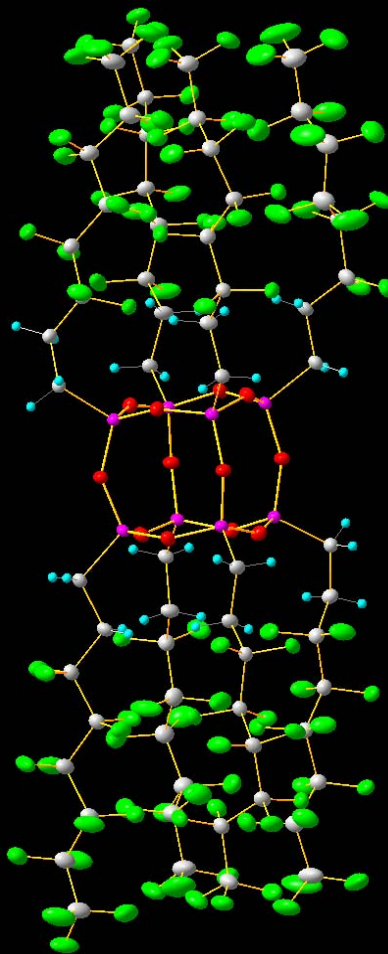




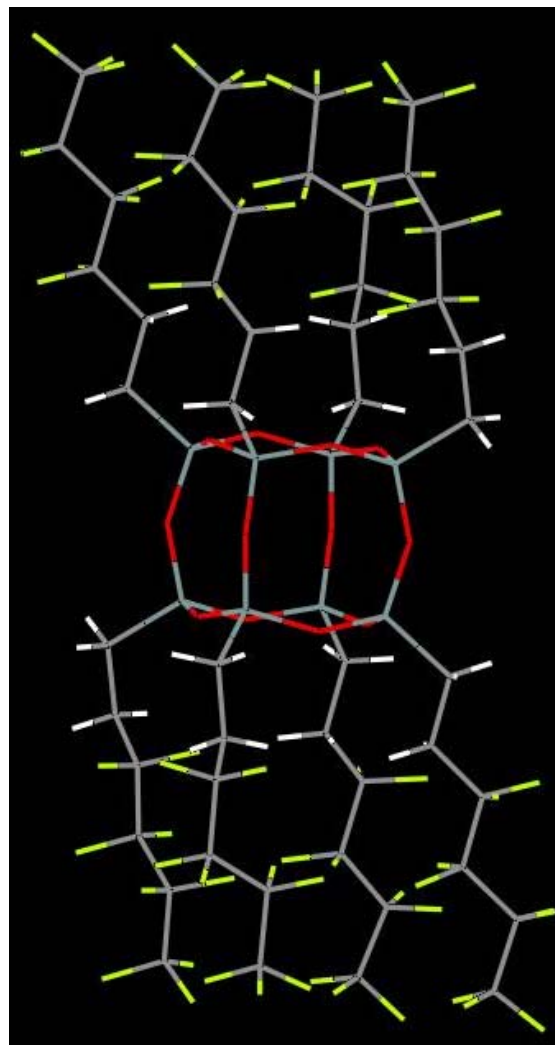
Crystal structures of FluoroPOSS₈T₈



Decyl-



Octyl-



Hexyl-

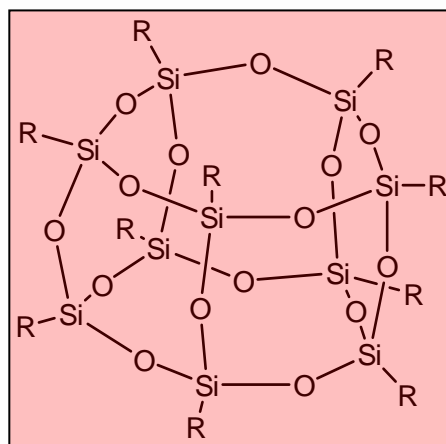
Distribution A: Approved for public release; distribution unlimited



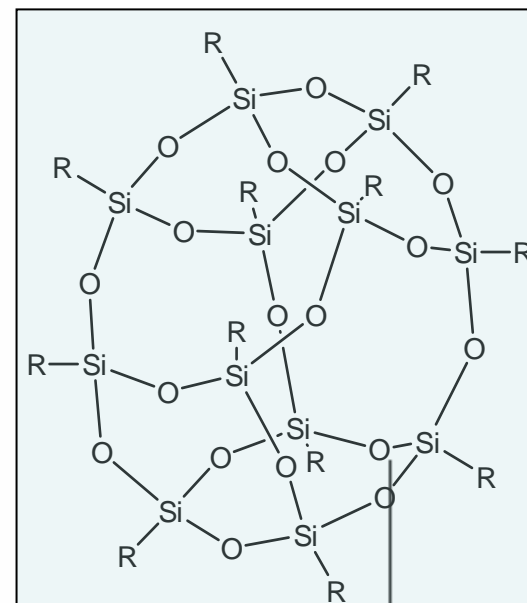
Fluoropropyl_nT_n

²⁹Si NMR

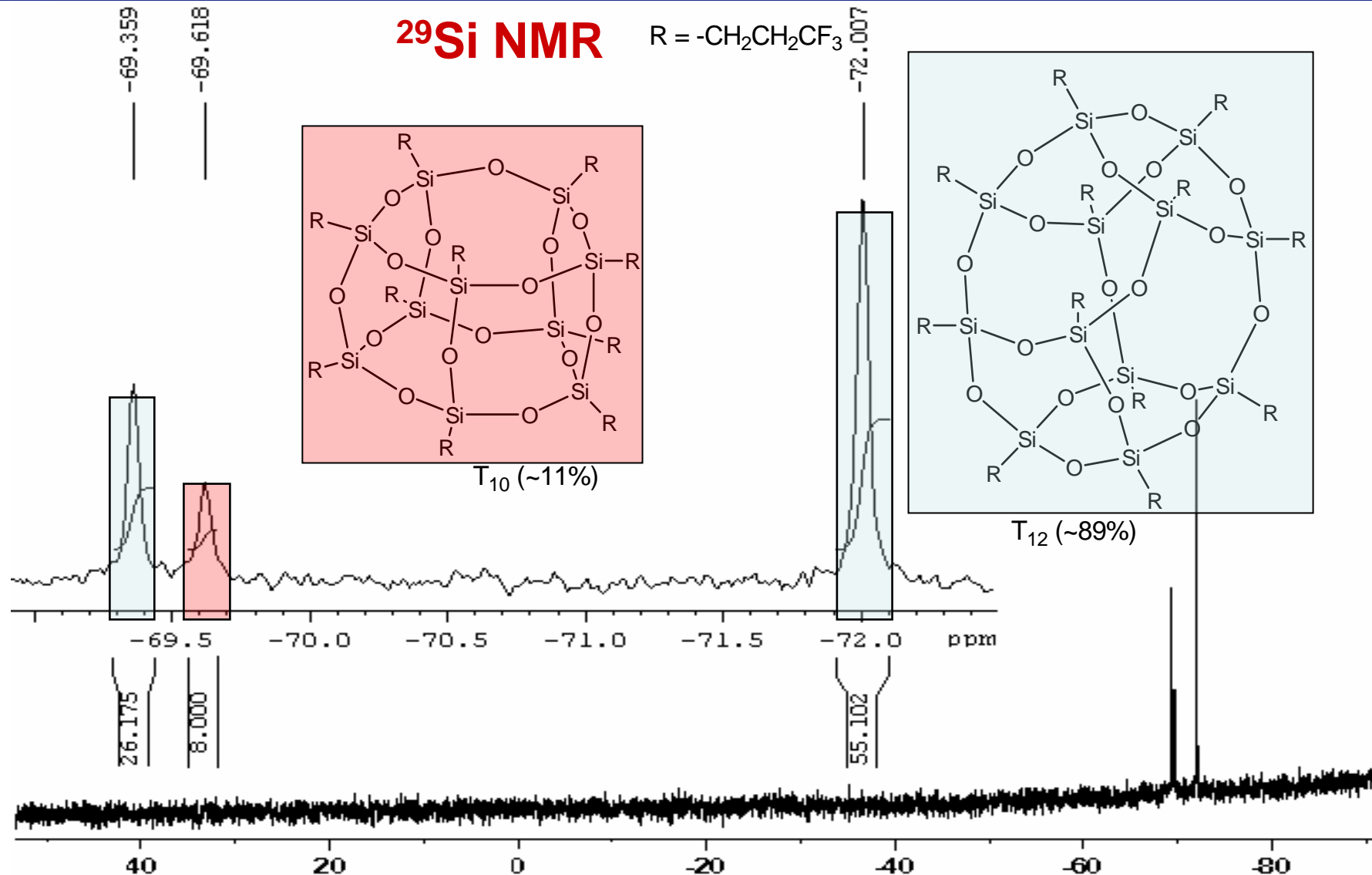
R = -CH₂CH₂CF₃



T₁₀ (~11%)

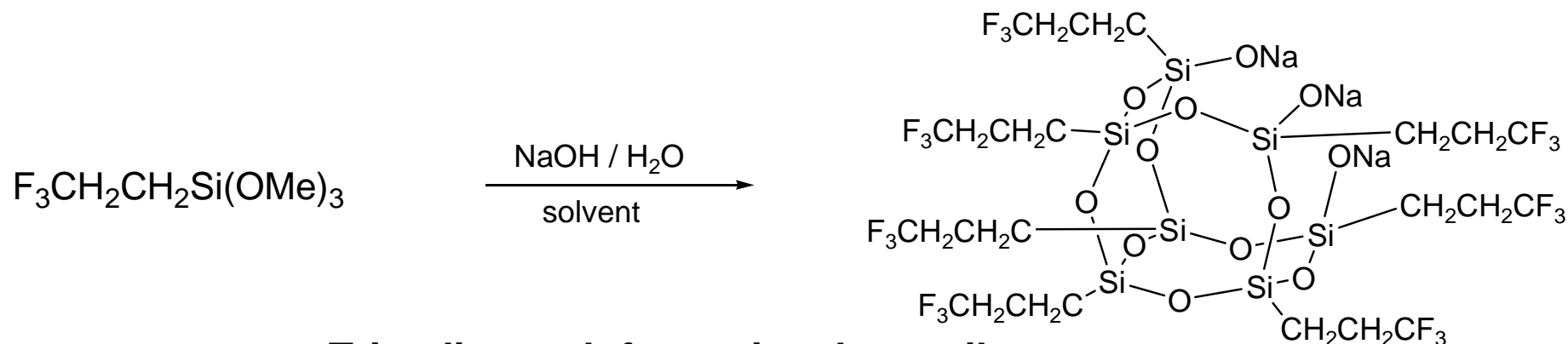


T₁₂ (~89%)

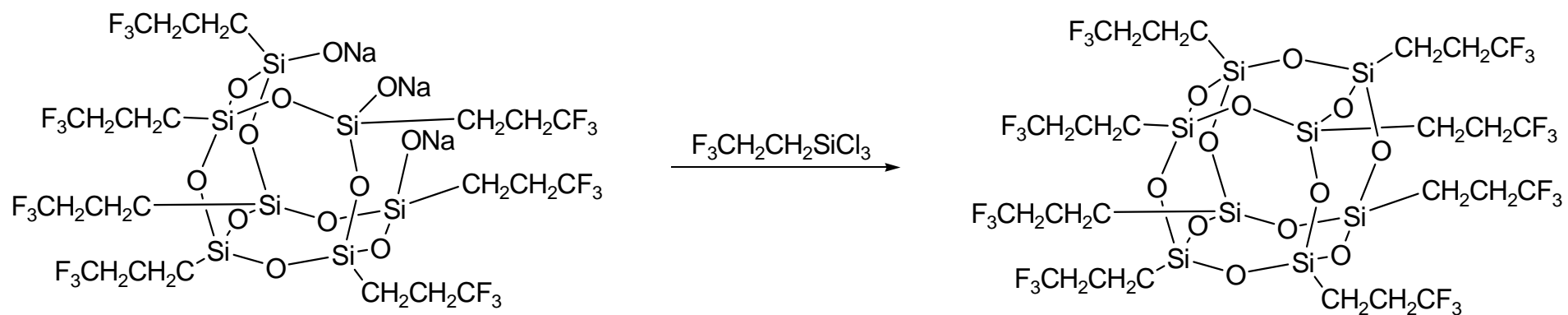




Fluoropropyl₈T₈



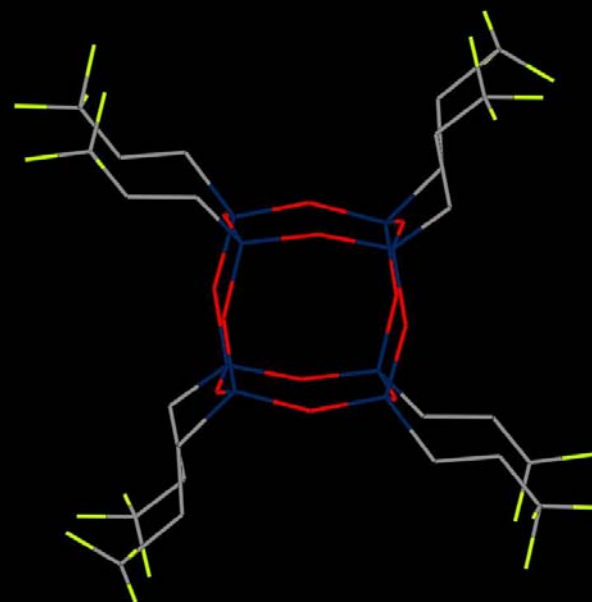
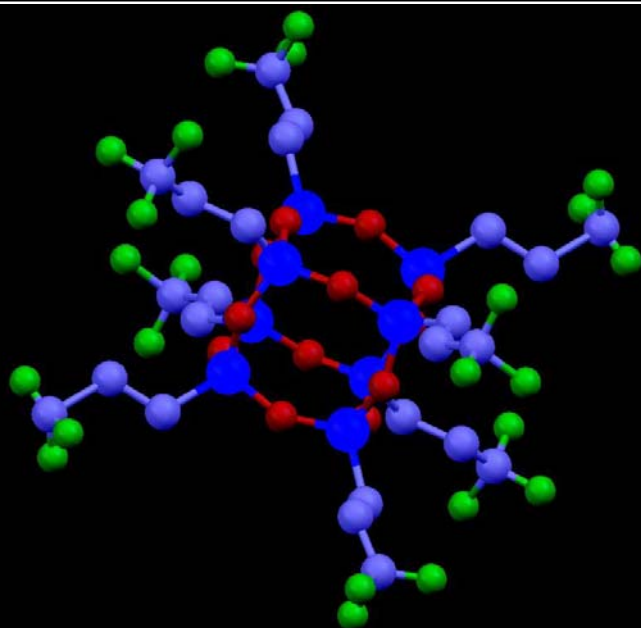
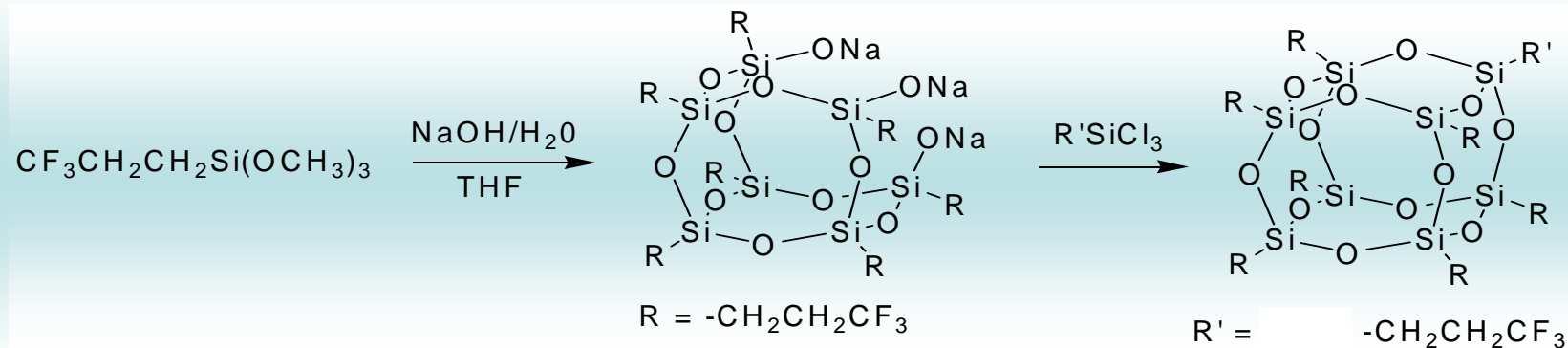
Trisodium salt from trimethoxy silane



T₈ from trisodium salt



Trifluoropropyl T_8 POSS





29Si NMR Trifluoropropyl POSS



Corner capped Trifluoropropyl POSS

Single peak in T₈ region

Current Data Parameters
NAME STID032
EXPNO 3
PROCNO 1

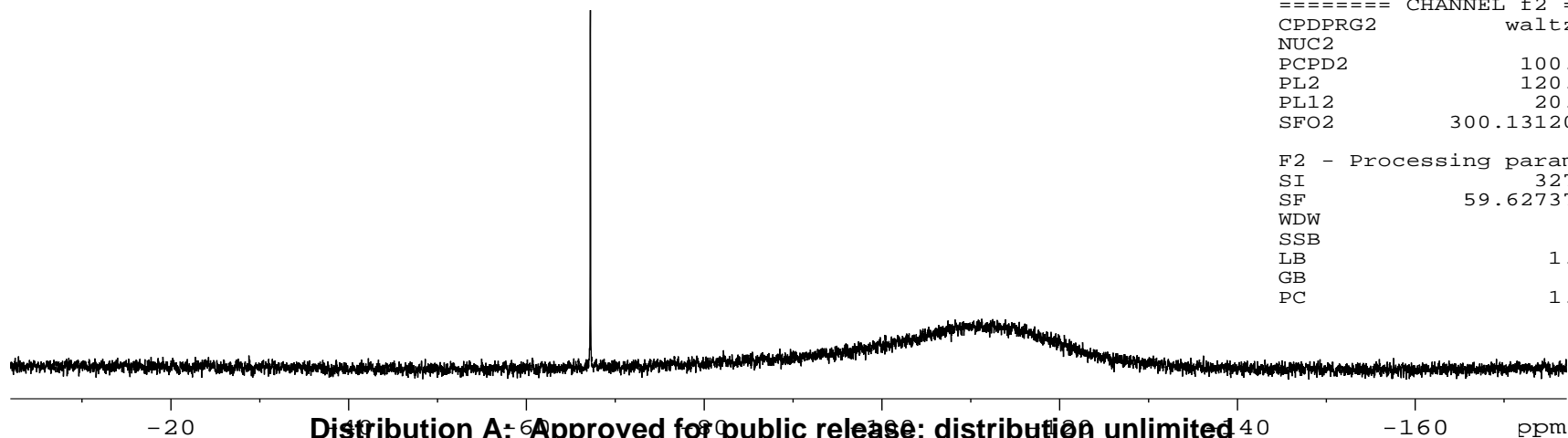
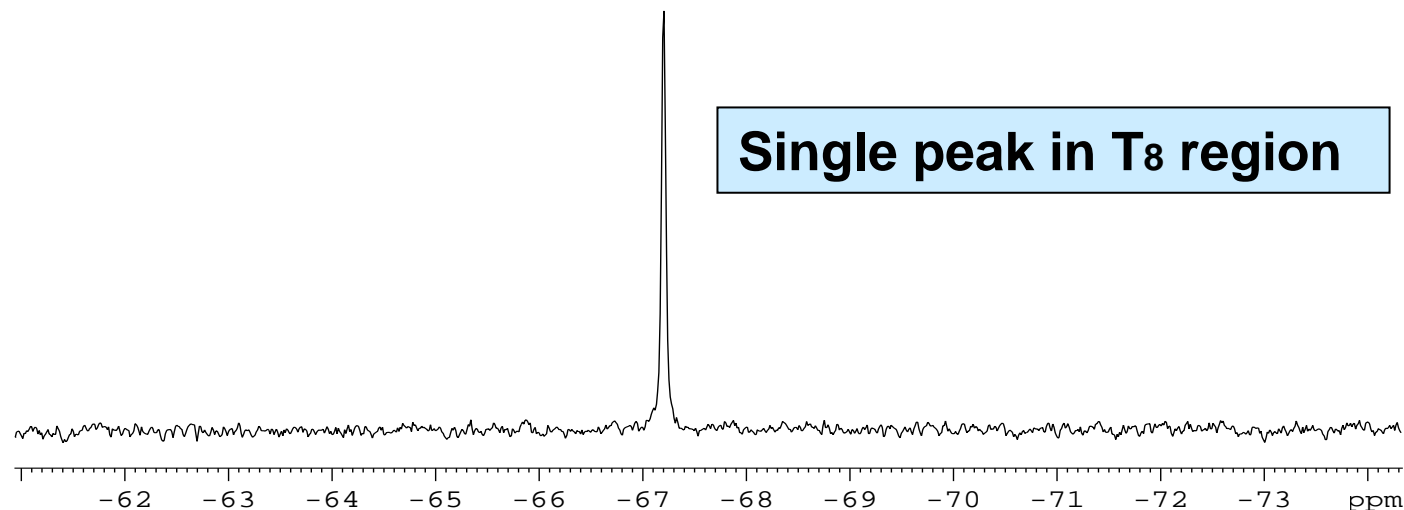
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SOLVENT CDC13
NS 993
DS 4
SWH 23809.523 Hz
FIDRES 0.363304 Hz
AQ 1.3763061 sec
RG 32768
DW 21.000 usec
DE 6.00 usec
TE 300.0 K
D1 8.00000000 sec
d11 0.03000000 sec

===== CHANNEL f1 =====
NUC1 29Si
P1 10.00 usec
PL1 -3.00 dB
SFO1 59.6214106 MHz

===== CHANNEL f2 =====
CPDPRG2 waltz16
NUC2 1H
PCPD2 100.00 usec
PL2 120.00 dB
PL12 20.00 dB
SFO2 300.1312005 MHz

F2 - Processing parameters
SI 32768
SF 59.6273782 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40



Distribution A: Approved for public release; distribution unlimited



1H NMR Trifluoropropyl POSS



Trifluoropropyl POSS

Current Data Parameters

NAME STID032
EXPNO 1
PROCNO 1

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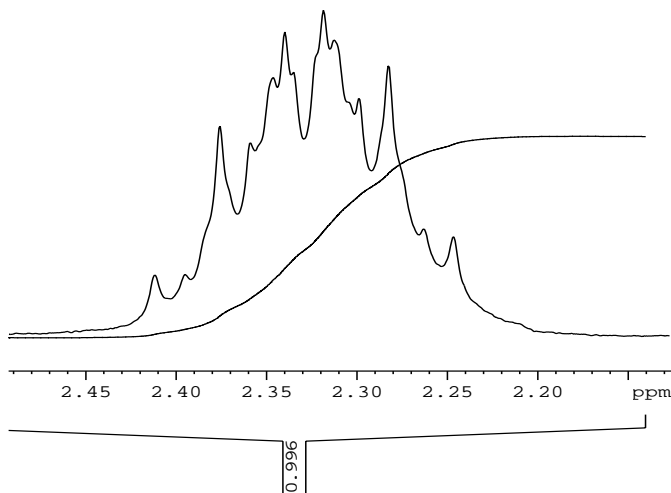
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PULPROG zg30
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SOLVENT CDCl3
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DS 2
SWH 6172.839 Hz
FIDRES 0.094190 Hz
AQ 5.3084660 sec
RG 4
DW 81.000 usec
DE 6.00 usec
TE 300.0 K
D1 1.00000000 sec

===== CHANNEL f1 =====

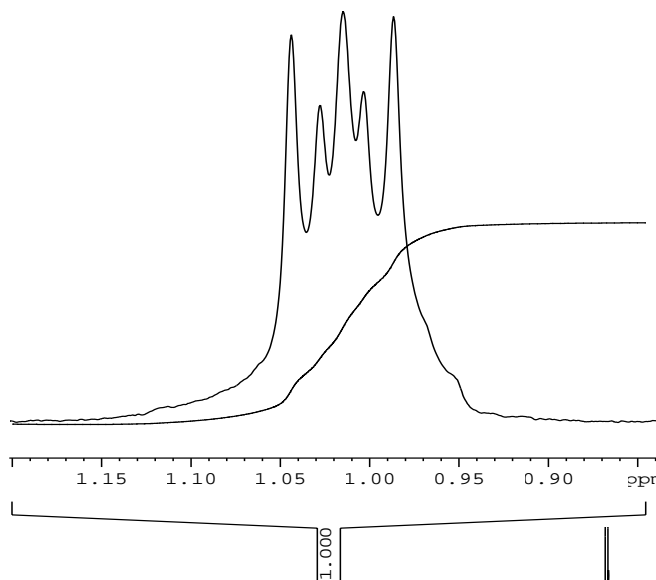
NUC1 1H
P1 9.50 usec
PL1 0.00 dB
SFO1 300.1318534 MHz

F2 - Processing parameters

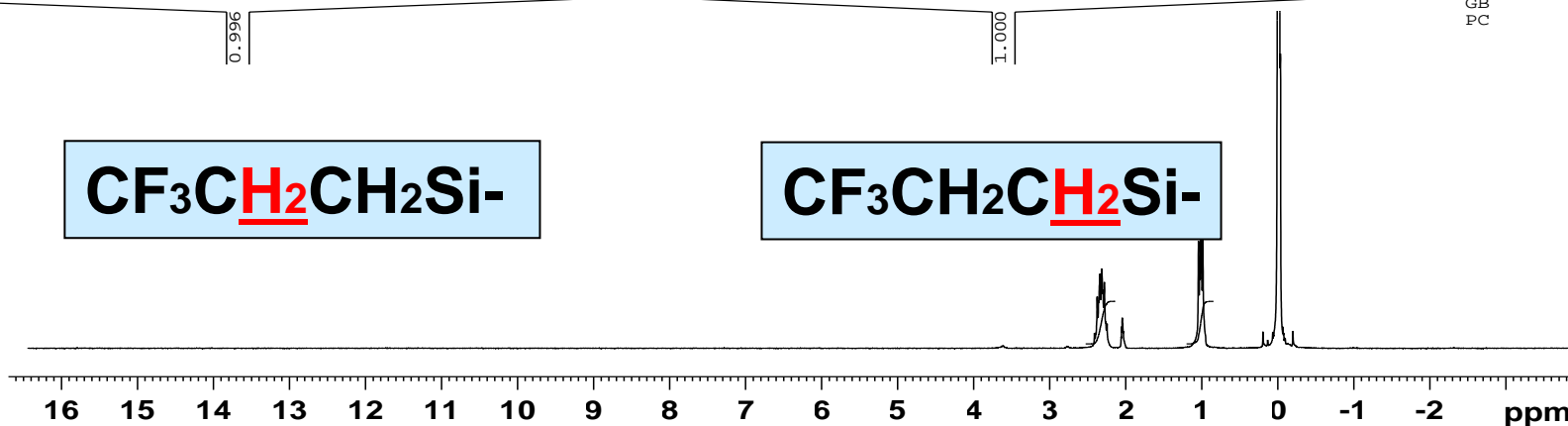
SI 32768
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WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00



CF₃CH₂CH₂Si-



CF₃CH₂CH₂Si-



Proton ratio 1:1



¹³C NMR Trifluoropropyl POSS



Corner Capped TFP-POSS

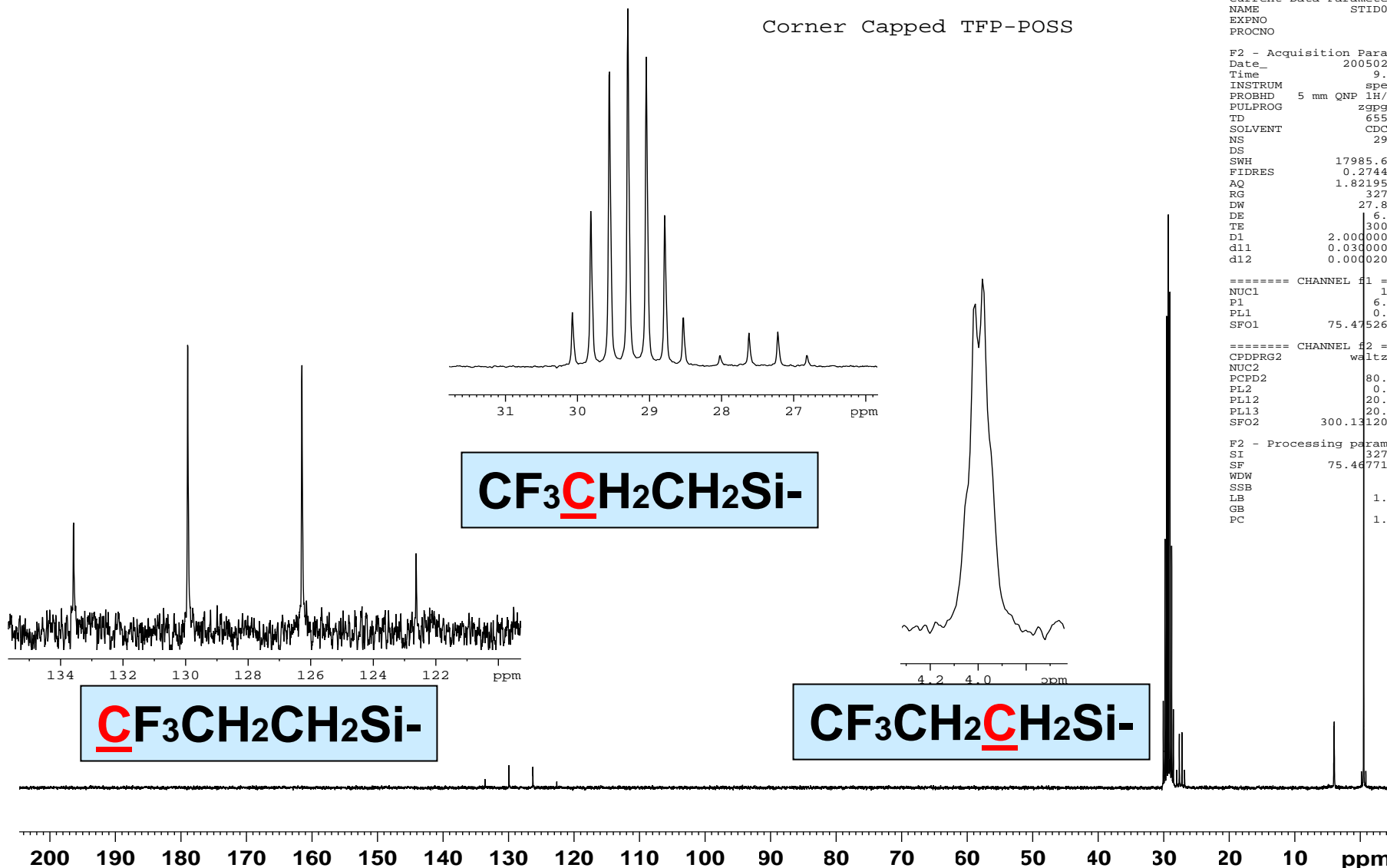
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EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
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TD 65536
SOLVENT CDCl3
NS 2982
DS 4
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FIDRES 0.274439 Hz
AQ 1.8219508 sec
RG 32768
DW 27.800 usec
DE 6.00 usec
TE 300.0 K
D1 2.00000000 sec
d11 0.03000000 sec
d12 0.00002000 sec

===== CHANNEL f1 =====
NUC1 13C
P1 6.25 usec
PL1 0.00 dB
SFO1 75.4752653 MHz

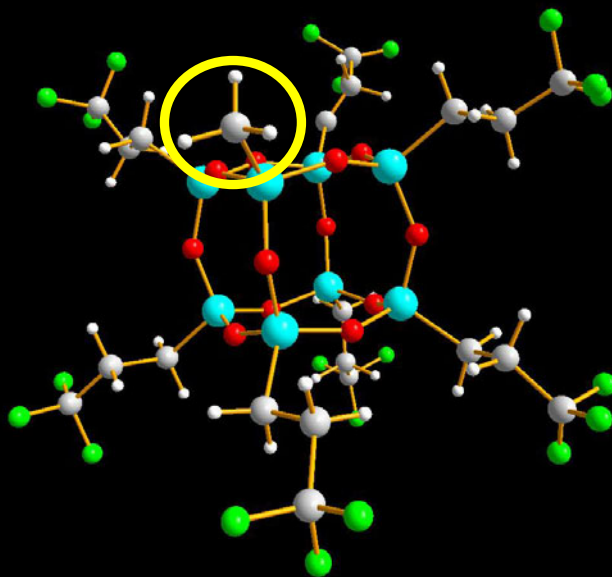
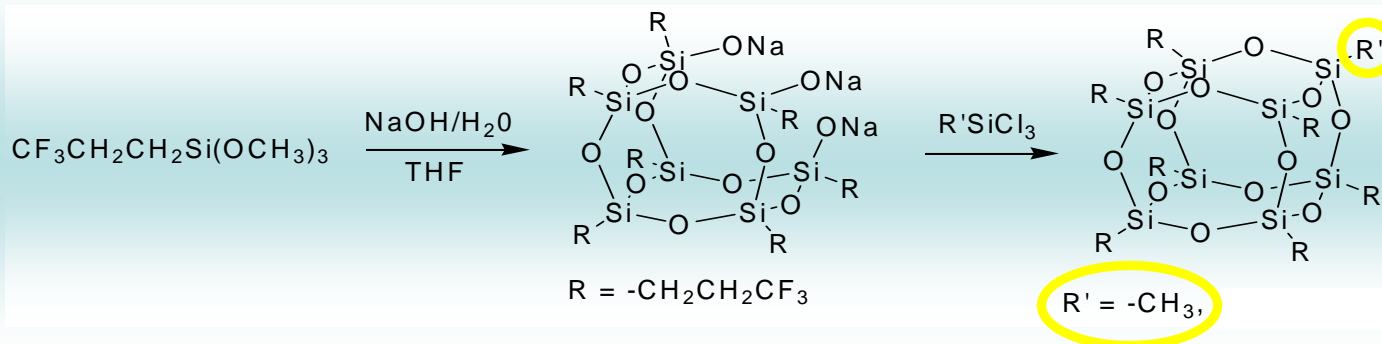
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CPDPRG2 waltz16
NUC2 1H
PCPD2 80.00 usec
PL2 0.00 dB
PL12 20.00 dB
PL13 20.00 dB
SFO2 300.1312005 MHz

F2 - Processing parameters
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WDW EM
SSB 0
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GB 0
PC 1.40



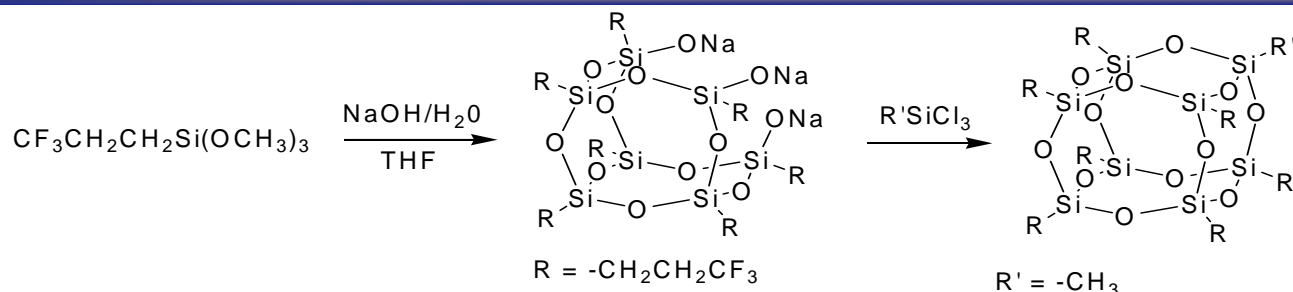


Me corner-capped Fluoropropyl T₈ POSS





Me corner-capped Fluoropropyl T₈ POSS

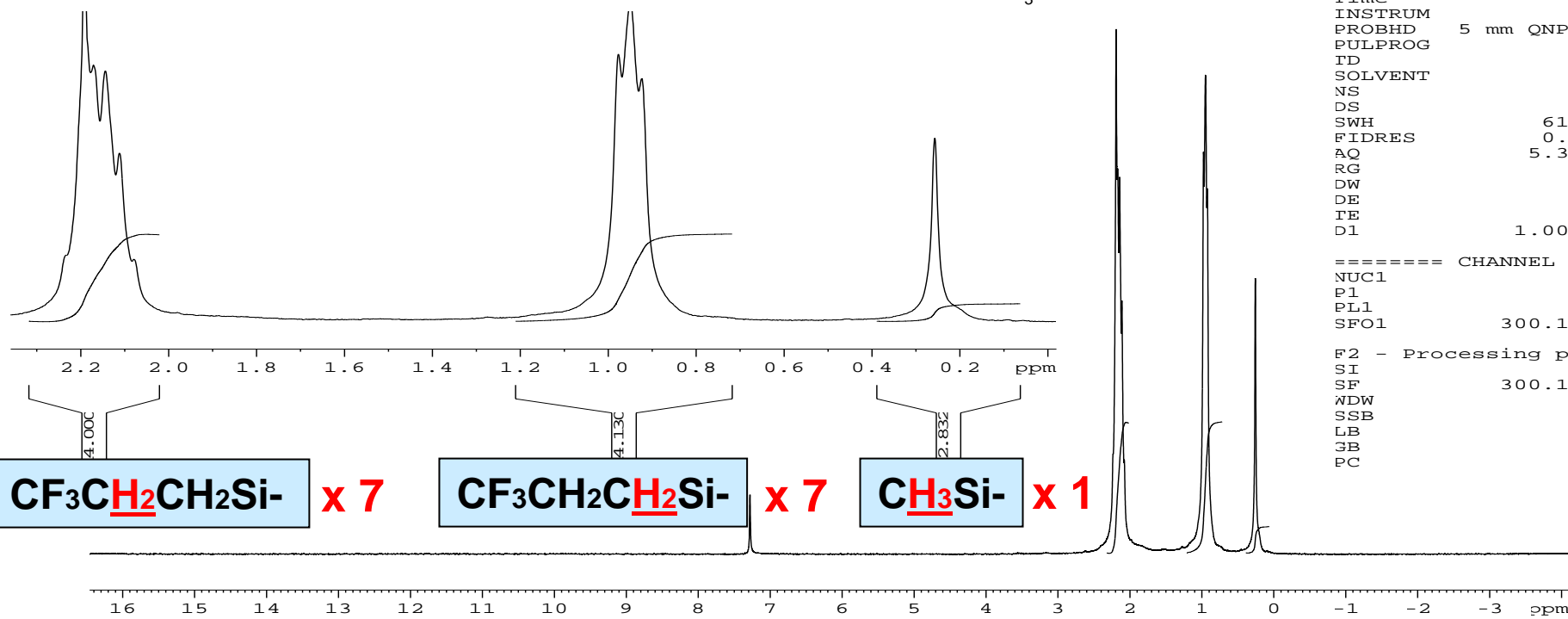


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EXPNO
PROCNO

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TD 655
SOLVENT CDC
NS
DS
SWH 6172.8
FIDRES 0.0941
AQ 5.30846
RG 456
DW 81.0
DE 6.
TE 300
D1 1.000000

===== CHANNEL f1 =
NUC1
P1 9.
PL1 0.
SFO1 300.13185

F2 - Processing parameters
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SSB
LB 0.
GB
PC 1.



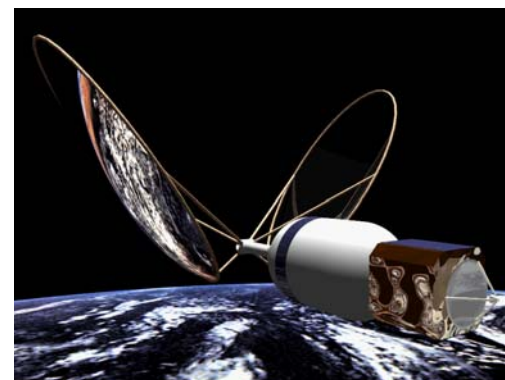
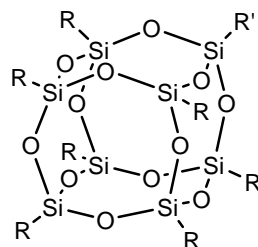
Proton ratio 14 : 14 : 3



Conclusions



- Conventional base catalyzed synthesis yields cage mixtures of fluoro-POSS
 - Redistribution method converts cage mixtures to pure T₈ POSS
- Fluoropropyl POSS does not readily convert to pure T₈ POSS as longer chained fluoro-POSS do
 - New synthetic corner-capping method produces pure T₈
- NMR and x-ray data determined that the new method was a success





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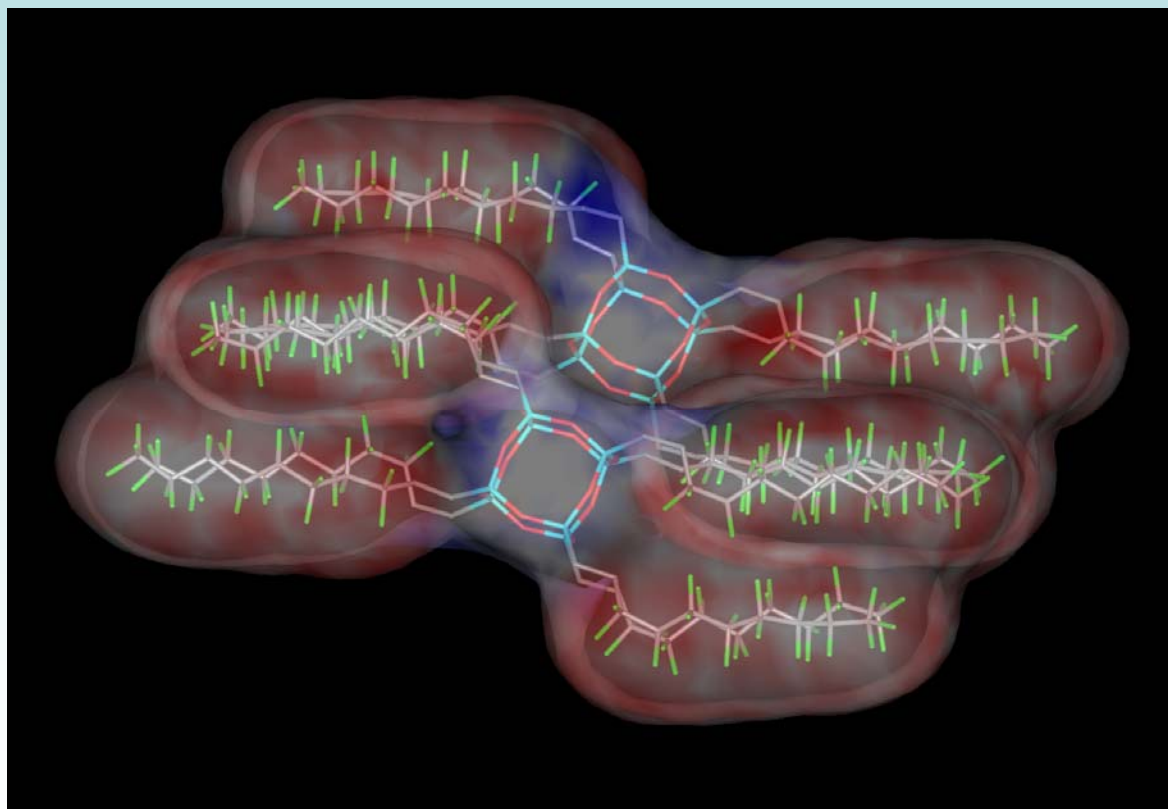


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Mrs. Sherly Largo
Dr. Tim Haddad
Lt. Will Cooper
Dr. Rusty Blanski
Lt. Laura Moody
Mrs. Sarah

AFOSR, PRS



AFRL Fluoropolymer Team

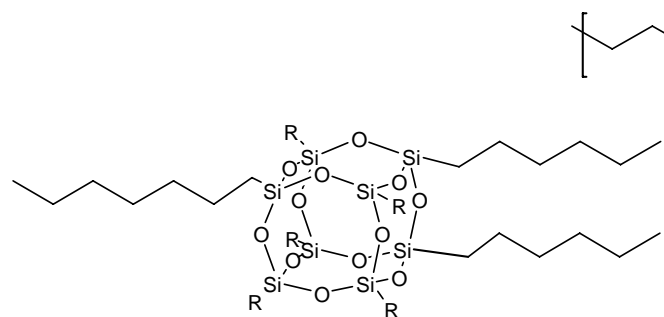


Joe Mabry
Ashwani Vij
Darrell Marchant
Scott Iacono
Tim Haddad
Laura Moody
Jerry Boatz
Pat Ruth
Isha Vij

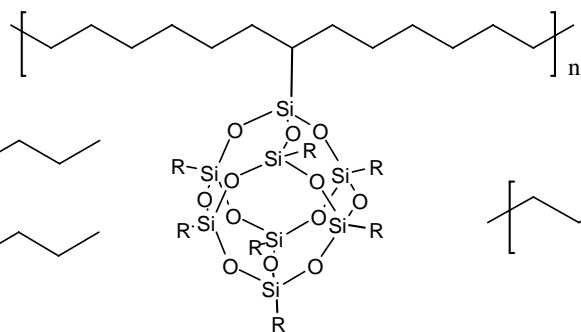
AFOSR
AFRL/PRSM
AFRL/PRSP



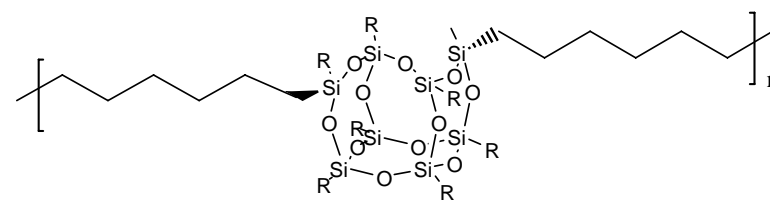
POSS polymer incorporation



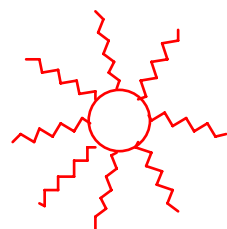
Cross-linker



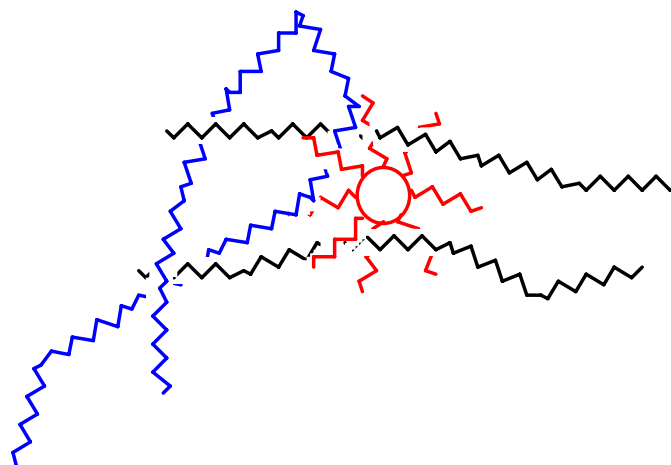
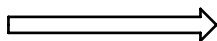
Pendant Polymer



Bead Copolymer

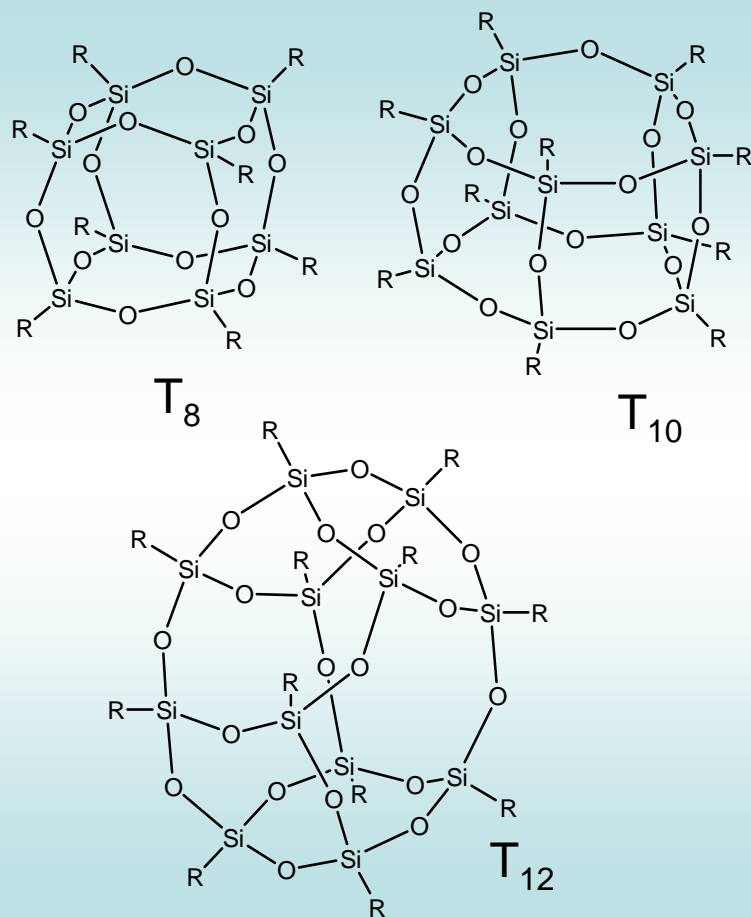


POSS Blending



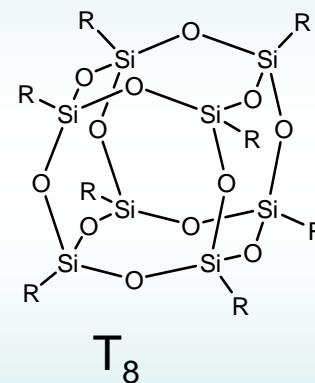


Redistribution reaction



**Sublimation
Re-crystallization
KOH Extraction**

→





Abstract



- Abstract

Fluorinated polyhedral oligomeric silsesquioxanes (POSS) are under investigation for potential lubricant applications. POSS compounds are characterized by a rigid, inorganic core consisting of silicon and oxygen, offering properties similar to those of ceramics; in addition to organic functional groups protruding from the inorganic core, offering organic functionality. Synthesis of POSS generally yields cage mixtures giving varying conformations of the POSS (Si-O) core denoted by the number of silicon atoms present with T8 being the most abundant conformation. Synthesis of trifluoropropyl-POSS is an exception to this generalization which adopts a preferred T10 conformation, with a minor amount of T8 and T12 isomers. Herein, we describe an alternate synthetic route for the preparation of pure T8 trifluoropropyl-POSS not possible by conventional methods. The POSS cages are characterized by multinuclear NMR as well as low-temperature, single crystal x-ray diffraction studies.